Subject: ENGG No. of CHEMISTRY No. of days/per Semester From date: 16.08.20 To Date: 11.12.2023	
JF	023
week class	
allotted: 04 No. of W	Veeks: 15
Week Class Day The	eory
1 ST Fundamental particles (electron, mass and charge)	proton & neutron Definition,
1 ST 2 ND Rutherford's Atomic model (pos	stulates and failure)
3 RD Atomic mass and mass number, I properties of Isotopes, isobars and	_
4 TH Bohr's Atomic model (Postulates	
1 ST Aufbau's principle, Hund's rule,	•
2 ND atomic no 30) Chemical Bonding: Definition NaCl, MgCl ₂	, Types, Electrovalent bond:
Covalent Bond wth examples I Coordinate bond NH ₄ +, SO ₂	H ₂ ,Cl ₂ , O ₂ , N ₂ , H ₂ O, CH ₄ , NH ₃ ,
4 TH Covalent Bond:H ₂ O, CH ₄ , NH ₃ ,	, Coordinate bond NH ₄ + , SO ₂
1 ST Concept of Arrhenius, Lowry Broand base with examples (Postula Neutralization of acid & base.	onsted and Lewis theory for acid
	pasic, double, complex and mixed s from each).
3 RD Definitions of atomic weight, mo weight	
4 TH Determination of equivalent weig	oht of Acid Base and Salt
1 ST Modes of expression of the conce & Molality) with Simple Problem	entrations (Molarity , Normality
2 ND pH of solution (definition with si	
4 TH Importance of pH in industry (su only)	•
Definition and types (Strong & weak) of Electrolytes with example.	
1 ST Electrolysis (Principle & process) and aqueous solution).) with example of NaCl (fused
2 ND Faraday's 1st law of Electrolysis	(Statement, mathematical
5 TH 3 RD Faraday's 1st law of Electrolysis (Statement, mathematical express numerical)	
4 TH Industrial application of Electroly	ysis- Electroplating (Zinc only)

	1 ST	Definition of Corrosion, Types of Corrosion- Atmospheric Corrosion
б ^{тн}	2 ND	Waterline corrosion. Mechanism of rusting of Iron only. Protection from Corrosion by (i) Alloying and (ii) Galvanization
	3 RD	Definition of Mineral, ores , gangue with example. Distinction between Ores And Minerals
	4 TH	i.Ore Dressing ii) Concentration (Gravity separation, magnetic separation)
7 ^{тн}	1 ST	Froth floatation & leaching
	2ND	Oxidation (Calcinations, Roasting)
	3RD	Reduction (Smelting, Definition & examples of flux, slag)
	4TH	Refining of the metal (Electro refining, & Distillation only)
8TH	1ST	Definition of alloy. Types of alloys (Ferro, Non Ferro & Amalgam) with example
	2ND	Composition and uses of Brass, Bronze, Alnico, Duralumin
	3RD	Saturated and Unsaturated Hydrocarbons (Definition with example)
	4TH	Alkane
	1ST	Alkene, Alkyne
9TH	2ND	Aliphatic and Aromatic Hydrocarbons (Huckle's rule only). Difference between Aliphatic and aromatic hydrocarbons
	3RD	IUPAC system of nomenclature of Alkane (up to 6 carbons) with bond line notation.
	4TH	IUPAC system of nomenclature of Alkene (up to 6 carbons) with bond line notation.
10ТН	1ST	IUPAC system of nomenclature of Alkyne (up to 6 carbons) with bond line notation.
	2ND	IUPAC system of nomenclature of alkyl halide and alcohol (up to 6 carbons) with bond line notation.
	3RD	Uses of some common aromatic compounds (Benzene, Toluene, BHC, Phenol) in daily life
	4TH	Uses of some common aromatic compounds (Naphthalene, Anthracene and Benzoic acid) in daily life.
	1ST	Sources of water
	2ND	Soft water, Hard water
	3RD	hardness, types of Hardness (temporary or carbonate and permanent or non-carbonate)

11TH	4TH	Removal of hardness by lime soda method (hot lime— Principle, process & advantages)
	1ST	Removal of hardness by lime soda method (cold lime— Principle, process & advantages) Advantages of Hot lime over cold lime process
12TH	2ND	Advantages of Hot lime over cold lime process
	3RD	Organic Ion exchange method (principle, process, and regeneration of exhausted resins).
	4TH	Definition of lubricant, Types (solid, liquid and semisolid with examples only)
13TH	1ST	specific uses of lubricants (Graphite, Oils, Grease), Purpose of lubrication
	2ND	Definition and classification of fuel, Definition of calorific value of fuel, Choice of good fuel
	3RD	Liquid: Diesel, Petrol, and Kerosene Composition and uses
	4TH	Gaseous: Producer gas and Water gas (Composition and uses).
	1ST	Elementary idea about LPG, CNG and coal gas (Composition and uses only).
	2ND	Definition of Monomer, Polymer, Homo-polymer, Co- polymer and Degree of polymerization
	3RD	Difference between Thermosetting and Thermoplastic
14TH	4TH	Composition and uses of Polythene, & Poly-Vinyl Chloride and Bakelite
	1ST	Definition of Elastomer (Rubber). Natural Rubber (it's draw backs)
	2ND	Vulcanisation of Rubber. Advantages of Vulcanised rubber over raw rubber
15TH	3RD	Pesticides: Insecticides, herbicides, fungicides Examples and uses.
	4TH	Bio Fertilizers: Definition, examples and uses.